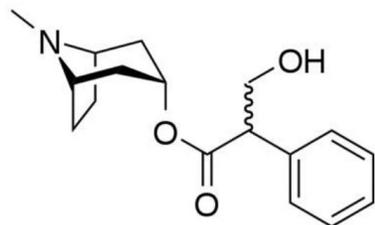
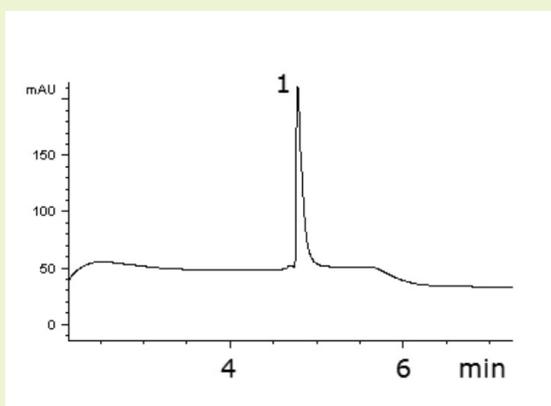


# Precise Determination of Atropine using a simple gradient

## No Ion Pair Reagent Necessary



**Atropine**



### Method Conditions

**Column:** Cogent Bidentate C18™, 4µm, 100Å

**Catalog No.:** 40018-75P

**Dimensions:** 4.6 x 75 mm

**Mobile Phase:** A: DI H<sub>2</sub>O + 0.1% acetic acid + 0.005% TFA

B: Acetonitrile + 0.1% acetic acid + 0.005% TFA

Both solutions were vacuum filtered through a 0.45µm nylon filter

Gradient:	time (min.)	%B
	0	10
	4	30
	6	30
	6.01	10

**Injection vol.:** 1µL

**Flow rate:** 1 mL/min

**Detection:** UV 214 nm

**Sample:** Prepared in 50% solution A/50% solution B, concentration 1 mg/mL and was filtered through a 0.45µm nylon membrane

**Peak:** 1. Atropine

Injection 1: RT = 4.772 min

Injection 2: RT = 4.773 min

Injection 3: RT = 4.772 min

Injection 4: RT = 4.774 min

### Discussion

Chromatographic separation and quantification methods of tropane alkaloids are often described in the literature and the method of choice is usually ion-pair chromatography (IPC), which requires long equilibration times and it is not very robust. This method shows a symmetrical peak for atropine using a simple gradient method that does not include any ion pair reagents which can cause damage to columns and lack reproducibility. The retention times are extremely repeatable but one of the best advantages to this method is the time savings between runs. Using a Cogent Bidentate C18 column the equilibration time was very short (5 min between every gradient run).

**Note:** After a simple sample clean up procedure, the method can be applied for monitoring atropine concentrations in biological specimens in cases of drug poisoning. The recoveries of atropine added to drug-free specimens which were analyzed using the described method were satisfactory with coefficients of variation of 4% or less.